

THERMAL-CATALYTIC ISOMERIZATION OF SUBSTITUTED VINYL CYCLOPROPANES

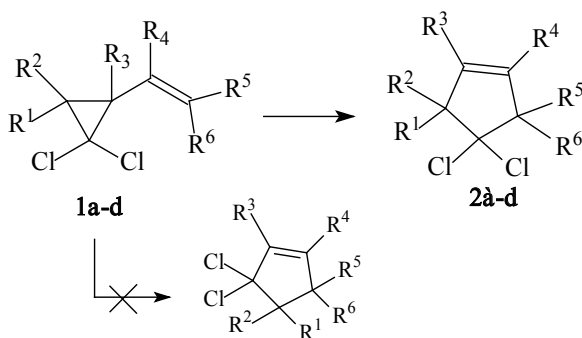
Raskildina G.Z.,¹ Borisova Yu.G.,¹ Davletshin A.R.,¹ Daminev R.R.,² Zlotskii S.S.¹

¹ Ufa State Petroleum Technological University, Ufa

² Sterlitamak branch of Ufa State Petroleum Technological University, Sterlitamak
e-mail: graskildina444@mail.ru

It is known that dichlorocarbene species of industrial dienes form vinyl-gem-dichlorocyclopropanes, which are widely used in the synthesis of reagents, oligo- and polymers¹.

We have carried out thermocatalytic isomerization of mono-dichlorocarbene olefins 1a-d to the corresponding gem-dichlorocyclopentenes 2a-d.



$R^1, R^2, R^3, R^4, R^5, R^6 = H$ (**1a**, **2a**), $R^1, R^2, R^3, R^5, R^6 = H, R^3 = CH_3$ (**1a**, **2a**),
 $R^1, R^2, R^4, R^5 = H, R^3, R^4 = CH_3$ (**1a**, **2a**), $R^3, R^4 = H, R^1, R^2, R^5, R^6 = CH_3$ (**1a**, **2a**),
 $R^1, R^2, R^4, R^5, R^6 = H, R^3 = Ph$ (**1a**, **2a**)

From literature² it follows that during thermal isomerization (450-550°C) vinyl-gem-dichlorocyclopropane undergo partial or complete dehydrochlorination. In our case (zeolites: SAPO-34, HY, ceocar-600; 280°C) 5-dichloro-isomeric isomeric cyclopentenes 2a-5 were not detected in the reaction products, which is explained by lower bond strength R_3C-CCl_2 (42-43 kcal / mol) than $R_3C-CR_1R_2$ (49-50 kcal / mol) in cyclopropane fragment³.

References

1. Kolesov S.V., Vorobyova A.I., Zlotskii S.S., Khamidullina A.R., Muslukhov R.R., Spirikhin L.V. // The Proceedings of the Academy of Sciences, 2008, 418, No. 2, p. 203-204 DOI: 10.1134/S0012500808010059.
2. Tomas Hudlicky and Josephine W. Reed // Angew. Chem. Int. Ed. 2010, 49, p. 4864 – 4876 DOI: 10.1002/anie.200906001.
3. Zhao Y., Truhlar D. G. // J. Chem. Phys. 2006, V. 125, Iss. 19, P. 194101. DOI: 10.1063/1.2370993.